ERSHLER, B.V.; BYAKOV, V.M.

Data on the recombination of radicals from various tracks in radiolysis, and homogeneous kinetics. Part 1. Setting the problem and calculation of certain parameters of a single spur according to the initial radiolysis yields. Zhur. fiz. khim.

36 no.4:913-915 Ap '62. (MIRA 15:6)

(Radicals (Chemistry)) (Photochemistry)

21.4100 5.4600 S/020/62/142/001/020/021 B145/B101

AUTHORS:

Firsov, V. G., and Ershler, B. V.

TITLE:

Slow reaction step in oxidations by hydrogen atoms according

to the Weiss mechanism

PERIODICAL: Akademiya nauk SSSR. Doklady, v. 142, no. 1, 1962, 145-147

TEXT: The effect of the concentration of added U(IV) and of the pH on the yields of radiolytic oxidation of aqueous Fe(II) and U(III) solutions by H atoms was investigated quantitatively to examine whether the theoretical predictions could be proved experimentally on the basis of the Weiss mechanism. Fe(II) is oxidized in the reactions: Fe(II) + OH \rightarrow Fe(III) + OH; Fe(II) + $H_2^{O_2} \rightarrow$ Fe(III) + OH \rightarrow He(III) + $H_2^{O_3} \rightarrow$ Fe(III) + $H_2^{O_4} \rightarrow$ Fe(I

Card 1/3

S/020/62/142/001/020/021 B145/B101 32821

Slow reaction step in oxidations ...

 $k_4[H^+]/k_6[U(IV)] = (\frac{1}{2} G - k)/(n + k - \frac{1}{2} G)$, (I), n and k being defined by equation (n + 2k) $H_2^0 \longrightarrow nH + mOH + kH_2 + 1H_2^0_2$. A comparison of the dependence of G on pH in U(IV) concentrations of 4.87.10-3 to 103.37.10-3 moles/liter with the curves calculated according to equation I showed good agreement. In this case, the Fe(II) concentration was $2 \cdot 10^{-3}$ -106·10⁻³ N. Co was used as source of radiation, the intensity of absorbed radiation being 1.60·10¹⁹ ev/liter·sec. In accordance with the theory, the curve has the same form as that obtained in a previous paper (DAN, 138, 1155 (1961)), in which $U0^{2+}$ instead of U(IV) was used as acceptor. According to the Weiss mechanism, the value of the quotient k_4/k_6 does not depend on the used active acceptors of the H_2^+ ion, which was proved experimentally when Fe(II) was replaced by U(III). The value k_4/k_6 calculated from the total yield $^G{\rm U}({\rm IV})$ + U(III) of the oxidation of U(IV) and U(III) was 1.35 in this case, whereas it was 1.25 when Fe(II)was used. For \sim 0.1 N U(IV) solution with an H⁺ concentration of \simeq 2, and with a content of $\sim 1\%$ of U(III), $G_{U(IV)} + U(III)$ was 8-8.2; this

Card 2/3

S/020/62/142/001/020/021 B145/B101

Slow reaction step in oxidations ...

corresponds to the G value of Fe(II) solutions. $G_{U(IV)} + U(III)$ does not change with increasing U(III) concentration. In a pure U(IV) solution, the slow step might be the reaction of uranium with the H_2^+ ions. There are 2 figures, 1 table, and 5 references: 4 Soviet and 1 non-Soviet. The reference to the English-language publication reads as follows:

The reference to the English-language publication loads as 15. T. Rigg, G. Stein, J. Weiss, Proc. Roy. Soc., A211, 375 (1952).

ASSOCIATION: Institut

Institut teoreticheskoy i eksperimental'noy fiziki Akademii nauk SSSR (Institute of Theoretical and Experimental Physics

of the Academy of Sciences USSR)

PRESENTED:

April 3, 1961, by A. N. Frumkin, Academician

SUBMITTED:

March 11, 1961

Card 3/3

5/020/62/144/005/015/017 B124/B138 Applicability of the equations of homoteneous kinetics to Ershler, B. V., and Byakov, V. II. radiolysis by high-intensity bombardments; Akademiya nauk SSSR. Doklady, v. 144, no. 2, 1962, 1103-1104 5.4600 AUTHORS: TEXT: An attempt is made to prove the validity of the dimplified model and others to describe the radiolysis of iron sulfate developed by Allen and others TXT: An attempt is made to prove the validity of the simplified model by all and others to describe the radiolysis of iron sulfate by a developed by Allen and others to describe to 1023 ev/cm. sec. The developed by Allen and others to describe the radiolysis of iron sulfate by developed by Allen and others to describe the radiolysis of iron sulfate by the radiolysis of iron s TITLE: PERIODICAL: constant vicins of n, m_2 and m_2 m_2 and m_2 equation $(2k+n)H_2O = (21+m)H_2O = kH_2+1H_2O_2+mOH+nH$. Here k, l, m, and n are equation $(2k+n)H_2O = (21+m)H_2O = kH_2+1H_2O_2+mOH+nH$. Obtained from water by an the quantities of these particles respectively, obtained from water by an the quantities of these particles respectively intensities the oxidation yield of absorbed power of 100 ev. At ordinary intensities the oxidation vield of the quantities of these particles respectively, obtained from water by an absorbed power of 100 ev. At ordinary intensities the oxidation yield of a absorbed power of 100 ev. At ordinary intensities are 100 ev. where FeSO golution is known to be G absorbed power of 100 ev. At ordinary intensities the oxidation yield it is a solution in the second solution is known to be G max and solution is known to be G max and solution is oxidized by all radicals according to the second solution is oxidized by all radicals according to the second solution is oxidized by all radicals according to the second solution is oxidized by all radicals according to the second solution is oxidized by all radicals according to the second solution is oxidized by all radicals according to the second solution is oxidized by all radicals according to the second solution is oxidized by all radicals according to the second solution is second solution.

5/020/62/144/005/015/017 B124/B138

Applicability of the equations ...

reactions: $2Fe^{2+} + H_2O_2 \longrightarrow 2Fe^{3+} + OH^-$ (I); $Fe^{2+} + OH \longrightarrow Fe^{5+} + OH^-$ (II); $H + O_2 \longrightarrow HO_2$ (III); $H + O_2 \longrightarrow HO_2$ (III); $H + O_3 \longrightarrow HO_3$ (III); $H + O_3$

Card 2/3

Applicability of the equations ...

S/020/62/144/005/015/017 B124/B138

$$k_3/k_5^{1/2} = (I/100N)^{1/2} \frac{n - 1/3(G_{max} - G)}{[(1/6)(G_{max} - G)]^{1/2}} \cdot [1/(O_2)]$$
where $G_{max} = 15.6$ (maximum yield of oxidized iron), k_3 and k_5 are the rate constants of the reactions (III) and (V) . Note that

where $G_{\rm max}$ = 15.6 (maximum yield of oxidized iron), k, and k, are the rate constants of the reactions (III) and (V), N is Avogadrots number, and (0_2) (0_2 concentration in the air-saturated solution) = $3.7 \cdot 10^{-4}$ mole/liter. The constant value of the $k_3/k_5^{1/2}$ constant confirms the equation given. There is 1 figure. The English-language references are: N. F. Barr, A. O. Allen, J. Phys. Chem. 63, 928 (1959); C. J. Hochanadel, J. Phys. Chem. 56, 587 (1952).

PRESENTED: January 14, 1962, by A. N. Frumkin, Academician

SUBMITTED: January 12,1962

Card 3/3

L 18876-63 EFF(c)/EWT(m)/BDS AFFTC/ASD Pr-4 WW/JFW ACCESSION NR: AP3006620 S/0076/63/037/009/2016/2021

AUTHORS: Byakov, V. M.; Ershler, B. V. (Moscow)

TITLE: Data on the recombination of radicals from different tracks during radiolysis and homogeneous kinetics.

4. Ranges of concentration and irradiation rates within which the simple radiolysis model is valid

SOURCE: Zh. fizicheskoy khimii, v. 37, no. 9, 1963, 2016-2021

TOPIC TAGS: radiation chemistry, irradiation, radical, radical theory, radical capture, radical recombination, radical diffusion, radiolysis, acceptor, oxidation, reduction, Allen model, track, gamma-ray, acceptor concentration, aqueous solution, I, fast electron, spur, electron volt, kinetics, homogeneous kinetics

ABSTRACT: Calculations are made which show that radiolysis of acqueous solutions by hard gamma-rays and fast electrons can be quantitatively described with a simplified model according to which radicals and molecular products are generated in constant yield throughout the entire volume of the solution. The existence of experimentally determinded I-correlation in the radiolysis of some aqueous

Card 1/2

L 18876-63

ACCESSION NR: AP3006620

solutions is proof that the active particles in such solutions are radicals rather than excited water molecules. Bibliographic reference to A. O. Allen (The Radiation Chemistry of Water and Aqueous Solutions, 1961). Orig. art. has: 4 formulas and 2 figures.

ASSOCIATION: none

SUBMITTED: 23Aug62

DATE ACQ: 30Sep63

ENCL: 00

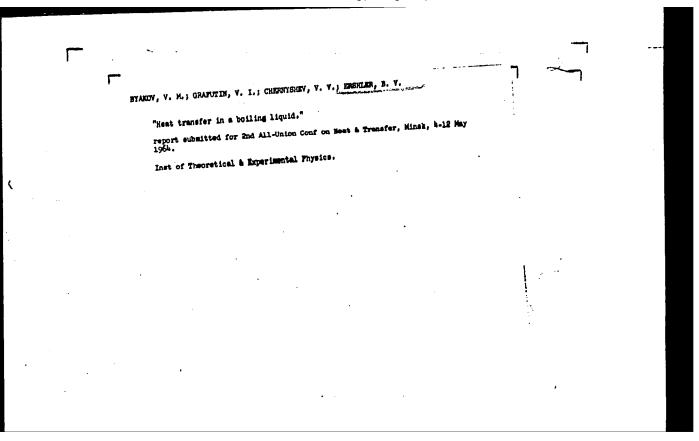
SUB CODE: CH, NS

NO REF SOV: 011

OTHER: OOL

2/2

Card



ACCESSION NR: AP4036526

AUTHOR: Memelova. L. Ya.; Kukawadze, G. M.; Ershler, B. V.

TITIE: Mass spectrometric determination of very small amounts of boron in certain materials

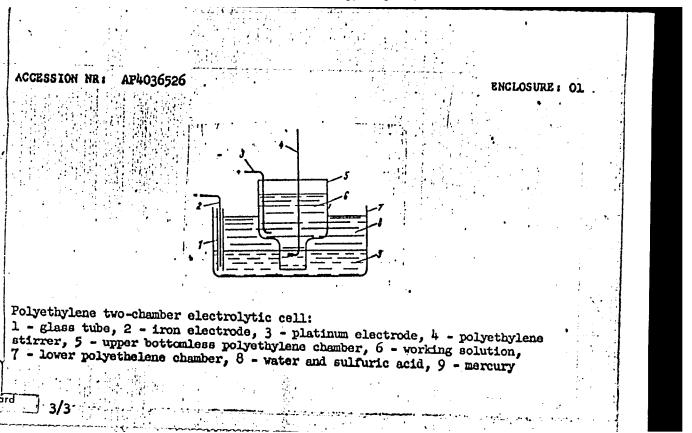
SOURCE: Atomnaya energiya, v. 16, no. 5, 1964, 423-426

TOFIC TAGS: boron determination, boron mass spectrometry, isotopic dilution method, analytical chemistry, boron, mass spectrometry

ABSTRACT: The method of isotopic dilution suggested by G. Morrison and R. Rupp (Analyt. Chem. 6, (1957), 892) was used for the determination of small amounts (Analyt. Chem. 6, (1957), 892) was used for the determination. The sample was of boron of the order of 10 fem in glass, quartz, and silicon. The sample was of boron of the order of 10 small endown the standard was added which condissolved (or fused) in sodium hydroxide, internal standard was added which condissolved for fused) in sodium hydroxide, internal standard was added which condissolved for fused in sodium hydroxide, internal standard was added which condissolved for fused in sodium hydroxide, internal standard was added which condissolved for fused in sodium hydroxide, internal standard was added which condissolved for fused in sodium hydroxide, internal standard was added which condissolved for fused in sodium hydroxide, internal standard was added which condissolved for fused in sodium hydroxide, internal standard was added which condissolved for fused in sodium hydroxide, internal standard was added which condissolved for fused in sodium hydroxide, internal standard was added which condissolved for fused in sodium hydroxide, internal standard was added which condissolved for fused in sodium hydroxide, internal standard was added which condissolved for fused in sodium hydroxide, internal standard was added which condissolved for fused in sodium hydroxide, internal standard was added which condissolved for fused in sodium hydroxide, internal standard was added which conditions and the fused in the f

"APPROVED FOR RELEASE: Thursday, July 27, 2000 CIA-RDP86-00513R00041222

Aturally occurring isotopic ratio. Orig. art. has: 1 figure SSOCIATION: None URMITTED: 22Jul63 DATE ACQ: 03Jun64 UB CODE: NP, GC NO REF BOV: CO3 OTHER		
SSOCIATION: None URMITTED: 22Jul63 DATE ACQ: 03Jun64 COTHER		
	.:.	
	R: 002	



FIRSOV, V.G.; KRSHLER, B.V.

Machanism of radiolysis in titanium salt solutions. Zhur. fiz. khim. 36 no.3:661-662 Mr '6?. (MIRA 17:8)

L 21333-65 EWT(m)/EWA(d)/EWP(t)/EWP(b) BSD/ASD(m)-3/IJP(c) MJW/JD/WB

ACCESSION NR: AP4044440 S/0076/64/038/008/1973/1977

AUTHOR: Vorob'yev, V. N. (Moscow); Ershler, B.V. (Moscow)

TITLE: Passivation of the titanium alloy AT-3 in solution of sulfuric acid

SOURCE: Zhurnal fizicheskoy khimii, v. 38, no. 8, 1964, 1973-1977

TOPIC TAGS: titanium alloy, passivation, AT 3, metal corrosion, sulfuric acid

ABSTRACT: An attempt is made to measure the capacity or the electric double layer during anodic oxidation of metal and thus to evaluate the amount of oxygen which is necessary to passivate AT-3 alloy. For the determination of the capacity of the double layer the active and reactive components of the electrode impedence of the double layer the active and reactive components of the electrode impedence were measured upon passage of small AC in a bridge circuit (see fig. 1 of enclosure), consisting of constant resistances R₂ and R₃, investigated cell 1, variable capacitor C₄ in series and variable resistance R₄. The investigated electrode A (0.8 m²) was made from sheet metal and placed inside the platinizes platinum A (0.8 m²) was made from sheet metal and placed inside the platinizes platinum coil. The capacitance of the electric double layer does not change over -0.42 v

ь 21333-65

ACCESSION NR: AP4044440

to +0.2 v potential interval, i. e. on passage from the active to the passive state. The results indicate that the amount of adsorbed oxygen on AT-3 titanium alloy in going from active to passive state is much less than that required for the formation of a monolayer. The thickness of the passivating oxide film does not exceed several A. Orig. art. has: 6 figures

ASSOCIATION: None

SUBMITTED: 04Sep63

ENCL: 01

SUB CODE: GC, MM

NR REF SOV: 015

OTHER: 006

Card / 3

L 21333-65 ACCESSION NR: AP4044440

ENCLOSURE: 01

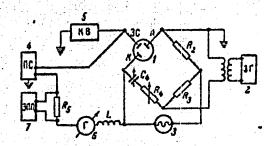


fig. 1
Capacitance measuring bridge circuit

1. cell; 2. pulse generator, 3. oscillograph, 4. potentiostat, 5. cathodic voltmeter, 6. galvanometer, 7. recording potentiometer

Card 3/3

Mechanism underlying the formation of molecular products in water radiolysis. Dokl. AN SSSR 154 no. 3:669-672 Ja '64. (MIRA 17:5)

1. Predstavleno akademikom A.I.Alikhanovym.

L 40882-66 E /T(1) WW/GD ACC NR AT6021837 (A) SOURCE CODE: UR/0000/65/000/000/0084/0099 Byskov. V. M.; Stepsnovs. O. P.; Ershler, B. V. ORG: Institute of Theoretical and Experimental Physics, Moscow (Institut teoreticheskoy i eksperimental noy fiziki) TITLE: Heat transfer and mixing in a boiling liquid SOURCE: Teplo- i massoperenos. t. III: Teplo- i massoperenos pri fezovykh prevrashcheniyakh (Heat and mass transfer. v. 3: Heat and mass transfer in phase transformations) Minsk, Nauks i tekhnika, 1965, 84-99 TOPIC TAGS: heat transfer, boiling, turbulent mixing ABSTRACT: The article first considers the growth of individual bubbles of vapor in a superheated liquid. If the radius, R, of the bubble is sufficiently great so that the capillary pressure due to the curvature of the surface can be neglected, the determining factor in the growth rate of a bubble of vapor is the rate of heat supply from the surrounding medium. The situation is described by the following equation: $\rho'' L \frac{d}{dt} \left(\frac{4\pi}{3} R^0 \right) = 4\pi R^0 \times (T' - T') \left(\frac{3}{\pi at} \right)^{1/a}$ (1)Card 1/2

L 40882-66

ACC NR: AT6021837

Here L is the heat of vapor formation; ρ " is the density of the vapor; T" is the temperature of the vapor; K and X are, respectively, the coefficients of the thermal conductivity of the liquid and the thermal diffusivity. Next follows a mathematical treatment of the case of the growth of a spherical bubble which is moving with respect to the liquid. Further sections of the article are devoted to a mathematical development for a vapor bubble in a boiling liquid, of turbulent diffusion in a boiling liquid, and a consideration of the laws governing the bubbling process. Orig. art. has: 26 formules and 6 figures.

SUB CODE: 20/ SUBM DATE: 09Dec65/ ORIG REF: 014/ OTH REF: 007

Card 2/2/17/2/

ERSUER, V.A., kand. tokhn. nauk, dotsent

Semiconductors for thermoelectric generators. Trudy LIVT no.71:54-56
164. (MIRA 18:10)

APPROVED FOR RELEASE: Thursday, July 27, 2000 CIA-RDP86-00513R000412220

ŧ

Using cables with aluminum cores on rivercraft. Rech. transp. 24 no.11:26-27 '65. (MIRA 19:1)

ERTE I.A., kandidat tekhnicheskikh nauk, dotsent; GUROVSKIY, N.Ya., inzhener; ERTE, I.A., inzhener.

Review of I.IA. IUkhim's and N.D.Zolotnitskii's books on safety engineering ("Safety measures in carpentry, stonemasonry and construction work." "Safety measures in pipe laying and plumbing." "Safety measures in mechanical woodworking, mechanical metalworking and forge work in building." I.IA. IUkhim. "Safety engineering in construction work." N.D.Zolotnitskii. Reviewed by B.V.Liul'ev. N.IA.Gurovskii. I.A.Erte). Gor.khos.Mosk. 24 no.2:44-46 F '50. (MIRA 7:11) (Building--Safety measures)

ERTEL, A. M.

Ertel, A. M. (USSR). (Magnetic Variometer). Russian Patent 126221, issued February 193h.

This invention relates to a magnetic variometer in which the effect of the magnetic field is increased by using iron rods fixed at both sides of the movable magnetic system of the variometer, the direction of the axis of the rods being along one straight line.

Claims allowed - 2.

ERTEL¹, A. M.

"The Hydro-Dynamic Theory of Lubrication on the Basis of the New Assumptions," Prik. mat. i mekh., 3, No.2, 1939

MRTEL!, A. H.,
"Theory of Semiliquid Friction" No. 9, 1944 Iz. Ak. Hauk. SSSR. otdel. Tekh. Lauk.
ER-52057019

METALI, A. H.	
"Hydrodynamic Conditions of Operating a Bearing in the Variable Process of Semiliquid Friction." No. 9, 1945. Iz. Ak. Nauk. SSSk. Othel. Tekn. nauk.	
DR-92059019	

ERTEL', G., Cand Chem Sci -- (diss) "Research into the melting of diaryliodo-salts." Moscow, 1960. 6 pp; (Moscow Order of Lenin and Order of Labor Red Banner State Univ im M. V. Lomonosov); 150 copies; price not given; (KL, 22-60, 133)

s/020/60/133/005/015/019 B016/B060

5.3200 AUTHORS:

Reutov, O. A., Corresponding Member AS USSR, Ertel'. G.

Ptitsina. O. A.

TITLE:

Reaction of Isotopic Exchange Between the Boron Fluorides

of Diaryl Iodonium and the Aryl Iodides Marked With 1131

PERIODICAL:

Doklady Akademii nauk SSSR, 1960, Vol. 133, No. 5,

pp. 1108-1110

TEXT: Basing on results obtained by A. N. Nesmeyanov (Refs. 1-5), L. G. Makarova and A. N. Nesmeyanov (Ref. 4), the authors arrived at the conclusion that the reaction mentioned in the title is possible; it was studied by them and can be represented as follows:

 $Ar_2I^+BF_4^- + Ar_2^{f_+BF_4^-} + Ar_2^{f_+BF_4^-} + Ar_2^{f_+BF_4^-}$

reagents required and of determining the activity are described. Table 1 gives the Ar2IBF4 synthesized by the authors for the first time, along with melting temperatures, analyses, and yields. The reaction of

Card 1/3

Reaction of Isotopic Exchange Between the Boron Fluorides of Diaryl Iodonium and the Aryl Iodides Marked With I¹³¹ S/020/60/133/005/015/019 B016/B060

isotopic exchange discussed here was found to take place only near the temperature at which diaryl iodonium boron fluoride decomposes. The authors studied the reactions of diphenyl iodonium boron fluoride with iodo benzene, and of di-(p-chlorophenyl)-iodonium boron fluoride with p-chloro-iodo benzene, of di-(p-bromophenyl)-iodonium boron fluoride with p-bromo-iodo benzene, of di-(p-tolyl)-iodonium boron fluoride with p-iodo toluene, of di-(p-anisyl)-iodonium boron fluoride with p-iodo anisole, of di-(m-carbethoxy-phenyl)-iodonium boron fluoride with m-iodo-benzoic acid ethyl ester, as well as of di-(m-nitro-phenyl)iodonium boron fluoride with m-iodo-nitro benzene. Table 2 gives the experimental conditions (temperature and period of reaction), moreover the degree of exchange and the diaryl iodonium boron fluoride yield. Because of varying thermal stability of the compounds, the experiments could not be carried out at the same temperature. Concerning the influence of the nature of the substituent in para- or meta-position of the benzene ring on the reaction rate of isotopic exchange it can therefore be only said that electronegative substituents delay the

Card 2/3

Reaction of Isotopic Exchange Between the Boron Fluorides of Diaryl Iodonium and the Aryl Iodides Marked With I¹³¹

S/020/60/133/005/015/019 B016/B060

reaction. This circumstance is apparently caused by the decrease of electron density on the iodine atom in the $\text{XC}_6\text{H}_4\text{I}$, with X being an electronegative substituent. In the case of non-substituted phenyl and in the monosubstituted benzene ring, diaryl iodonium boron fluorides form with a sufficiently high activity. The reaction of isotopic exchange can be recommended as a convenient method of producing IJ31-marked diaryl iodonium salts. There are 2 tables and 5 references: 4 Soviet and 1 German.

ASSOCIATION:

Moskovskiy gosudarstvennyy universitet im. M. V.

Lomonosova (Moscow State University imeni M. V. Lomonosov)

SUBMITTED:

May 6, 1960

Card 3/3

S/062/61/000/002/003/012 B115/B207

5 3700

2209

AUTHORS:

Ptitsyna, O. A., Reutov, O. A., and Ertel', G.

TITLE:

Synthesis of organoantimony compounds by means of

diaryl iodonium salts

PERIODICAL:

Izvestiya Akademii nauk SSSR. Otdeleniye khimicheskikh

nauk, no. 2, 1961, 265-270

TEXT: To synthesize organoantimony compounds, the authors used iodonium selts, a method which they had described earlier. To determine the optimum conditions, they decomposed both the double salt of diphenyl iodonium chloride and of antimony trichloride, as well as a mixture of $(C_6H_5)_2ICl$ and $SbCl_3$. The double salt $[(C_6H_5)_2ICl]_2 \cdot SbCl_3$ was prepared by pouring together the solutions of diphenyl iodonium chloride and of antimony trichloride. Independently of the ratio of initial components, it had the same composition. The double salt $[(C_6H_5)_2ICl]_2 \cdot SbCl_3$ was decomposed with antimony powder in acetone or ethyl acetate at different temperatures and different ratios of reagents.

Card 1/6

S/062/61/000/002/003/012 B115/B207

Synthesis of organoantimony ...

The following table shows the results of this study:

Molecular ratio of the double salt and antimony	Reaction conditions	Time of reaction, in hr	Total yield of organoantimony compounds, in %	\/
1:1.6	in acetone at room			$-\nu$
, , , , ,	temperature	25	15	
1 : 4.8	#	55	33	
1:3.0	in boiling acetone	3	63	
1:4.8	in ethyl acetate at			
	room temperature	55	38	
1:4.8	in boiling ethyl			
•	acetate	6	22	
1:4.8	11	l 12 i	27	

The yield was referred to the iodonium double salt used for the reaction. The table shows that boiling acetone is the best solvent. A mixture of organoantimony compounds:

Card 2/7/2

Synthesis of organoantimony ...

S/062/61/000/002/003/012 B115/B207

$$\begin{split} 9 & [(C_0H_0)_2 \text{NCI}]_2 \cdot \text{SbCI}_3 + 8\text{Sb} \rightarrow 6 & (C_0H_0)_3 \text{SbCI}_3 + 18C_0H_0J + 11\text{SbCI}_3 \\ 3 & [(C_0H_0)_2 \text{JCI}]_2 \cdot \text{SbCI}_3 + 2\text{Sb} \rightarrow 3 & (C_0H_0)_2 \text{SbCI}_3 + 6C_0H_0J - 2\text{SbCI}_3 \\ 3 & [(C_0H_0)_2 \text{JCI}]_2 \cdot \text{SbCI}_3 + 4\text{Sb} \rightarrow 3 & (C_0H_0)_2 \text{SbCI} + 6C_0H_0J + 4\text{SbCI}_3 \\ \end{split}$$

forms in the decomposition of the iodonium double salts by antimony powder just as in the decomposition of the double diazonium salts. The above-described method has the disadvantage that it is based on the preparation of iodonium double salts. The authors proved, however, experimentally in the present study that in the decomposition of mixtures $(c_6H_5)_2ICl$ and $sbcl_3$ with antimony powder in boiling acetone, the organoantimony compounds formed with the same yield as in the decomposition of the double salt $(c_6H_5)_2Icl)_2\cdot sbcl_3$. The decomposition of the Ar₂ICl and $sbcl_3$ mixture by antimony in boiling acetone was further applied by the authors to the synthesis of organoantimony

S/062/61/000/002/003/012 B115/B207

Synthesis of organoantimony ...

compounds with other radicals. As it may be seen from Table 2, the diaryl iodonium chloride decomposition with antimony in the presence of SbCl3 leads to mixtures of organoantimony compounds in satisfactory yield (60-70%). In this case, the organoantimony compounds are obtained by subjecting the diaryl iodonium double salts of antimony trichloride to decomposition. The authors base this assumption on the fact that when boiling the mixture of $(C_6H_5)_2ICl$ and SbCl₃ in acetone in the absence of antimony powder, they obtained only a double salt only from the acetone solution of the following composition (C6H5)2ICl 1.8 SbCl. The salt obtained is very likely a mixture of two iodonium double salts $(c_6H_5)_2Icl \cdot Sbcl_3$ and $[(c_6H_5)_2Icl]_2 \cdot Sbcl_3$. Furthermore, the authors answered the question as to whether organometal compounds of antimony may be obtained by the interaction of diaryl iodonium chloride with antimony powder in the absence of antimony trichloride. They found that organoantimony compounds formed in the reaction of diphenyl iodonium chloride with antimony in acetone, that

89907 \$\,\062\/61\/000\/002\/003\/012 \$\,\015\/\03207

Synthesis of organoantimony ...

the yield, however, depended on the kind of antimony used. Thus, the yield was 28% when a fine-ground powder of metallic antimony was used. It was 24% when antimony prepared in the following way was used: An equimolar quantity of zinc powder was admixed to the solution of SbCl in dilute hydrochloric acid; the resulting precipitate was filtered off, washed out with dilute hydrochloric acid, water, and a sodium carbonate solution (to remove HCl traces), then washed again with water, alcohol, and ether, and dried in air. The yield in organoantimony compounds was 50% if the antimony thus prepared was used (but without sodium carbonate treatment). When decomposing $(C_6H_5)_2$ ICl with fine-

ground antimony powder in the presence of hydrochloric acid traces, the authors obtained a 52% yield. On the basis of these data, they finally state that hydrochloric acid promotes the formation of organoantimony compounds in the decomposition of $(c_6H_5)_2$ ICl with antimony. From the

mentioned data it may be seen that the decomposition of diphenyl iodonium chloride with antimony leads to the formation of organoantimony compounds. Thus, antimony trichloride favors this reaction. When diphenyl iodonium iodide was used instead of diphenyl iodonium chloride, the yield

Card 5/6

89907

Synthesis of organoantimony ...

S/062/61/000/002/003/012 B115/B207

in organoantimony compounds fell considerably (9% as compared to 50%). There are 2 tables and 15 references: 7 Soviet-bloc and 8 non-Soviet-bloc.

ASSOCIATION:

Moskovskiy gosudarstvennyy universitet im. M. V. Lomonosova

(Moscow State University imeni M. V. Lomonosov)

SUBMITTED:

November 26, 1959

Card 6/7/2

PTITSYNA, O.A.; KOZLOVA, A.N.; REUTOV, O.A.; Prinimal uchastiye ERTEL', G., aspirant

> Synthesis of organoantimony compounds via double diaryliodonium salts of antimony pentachloride. Izv.AN SSSR Otd.khim.nauk no.4: 634-638 Ap 162. (MIRA 15:4)

1. Moskovskiy gosudarstvennyy universitet im. M.V.Lomonosova. (Antimony organic compounds) (Iodonium compounds)
(Antimony chlorides)

ERTEL, Gunter, inz.

Standardization and further development of long-distance belt conveying. Uhli 7 no.2:71-76 '65.

1. Institute of Conveying Engineering, Leipzig.

"APPROVED FOR RELEASE: Thursday, July 27, 2000

CIA-RDP86-00513R00041222

EWT(d)/FCC L 31368-66 IJP(c) ACC NR: AP6021119' SOURCE CODE: HU/0033/65/069/04-/0193/0197 AUTHOR: Cadez, Harjan (Professor; Doctor; Belgrade); Ertel, Hans (Professor; Doctor; Berlin) ORG: /Cadez/ Institute for Meteorology, University of Belgrade, Belgrade; /Ertel/ B Institute for Physical Hydrography, DAW, Berlin (Institut fur Physikalische Hydrographie der Deutschen Abederie der Water der Deutschen Abederie der Deutsche der Deutschen Abederie der Deutsche der phie der Deutschen Akademie der Wissenschaften) TITIE: Analogous characterizations of the continuity equation SOURCE: Idojaras, v. 69, no. 4-5, 1965, 193-197 TOPIC TAGS: Lagrange equation, partial derivative ABSTRACT: Analogous characterization of the continuity equation was accomplished by combining the local differentiation with respect to time and the Lagrangian equations for continuity. The analogous characterization thus obtained represents material variables, as verified by the rule for the differentiation of determinants together with the commutative property of partial derivatives. Orig. art. has: 22 formulas. [Orig. art. in German] [JPRS]

SUB CODE: 12 / SUBM DATE: none / ORIG REF: 004 / OTH REF: 006

Card 1/1 (C

35969-60	<u> FCC</u>							
ACC NR:	AP602738	6		SOURCE CO	DE: HU	/0033/65	/069/006/0	327/0331
AUTHOR: Borlin)	Cadoz, N	larjan (Pro	fessor; Docto	r; Belgrade); Erto	l, Hans	(Professor	: Doctor:
	adoz/ Fac	ulty of App	lied Nathemat	ics. Instit	ite of	Metoorol	ogy. Univo	rsity
of Bolgr loski Za	vod); Er	rado (Pirod: tel/ I <u>nstit</u> :	no-Hatomatick uto for Physi Hydrographio	i Fakultat cal Hydrogr	Univerz aphy. D	iteta u AV. Borl	Beogradu.	Kotooro-
TITLE:	Sedimonta urbulent	tion in the diffusion s	atmosphere u tream	ndor porsis	tonce o	f the in	trinsic so	ources B
SOURCE:	Idojaras	, v. 69, no	. 6, 1965, 32	7 - 331				
TOPIC TA	GS: atmo	spheric dif	fusion, moteo	rology				
ground 1 intrinsi of the c	evol was c sourcos onsiderat	investigatod of the turk ions discuss	on of suspond by introduc oulont diffus sed correlate German.	ing the printion flow. ' d well with	nciple The dat observ	of porsi	stence for ated on th	tho hasis
SUB CODE	: 04 /	SUBM DATE:	none / OR	IG REF: 000	6 / o	TH REF:	002	
ns Card 1/	1					-		

"APPROVED FOR RELEASE: Thursday, July 27, 2000

CIA-RDP86-00513R00041222

USSR/Atomic and Molecular Physics - Physics of the Molecule D-2

Abs Jour : Ref Zhur - Fizika, No 1, 1958, 698

Author : Laya, M.V., Ertelis, U.Ya.

Inst : -

Title : Concerning the Thermal Stability of Complex Molecules.

Orig Pub : Uch. zap. Rizhsk. ped. in-t, Raksti Rigas ped. inst.,

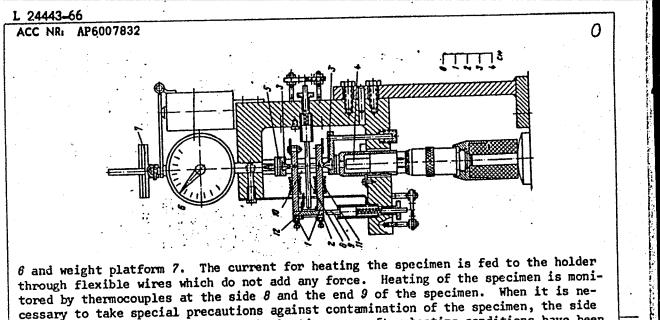
1957, 4, 151-156

Abstract : The authors compute the decay probability relative to the

valent bond of a linear three-atom molecule. Unlike the wirk by I.V. Obreimov (Zh eksperim i teor fiziki, 1946, 19, 396), account is taken of the anharmonic nature of the oscillations. The sought probability turns out to be dependent on the vibrational energy at a given bond and on the anharmonicity constant, and increases with the

latter.

Card 1/1



through flexible wires which do not add any force. Heating of the specimen is monitored by thermocouples at the side θ and the end θ of the specimen. When it is necessary to take special precautions against contamination of the specimen, the side thermocouple may be removed from the heating zone after heating conditions have been determined for a whole series of identical specimens. While the specimen is being heated, the top holder moves freely upward along guides 10. Heating elements 11 may be used for more uniform heating of the specimen. Resistance oven 12 has the same

2/3

L 24443-66 ACC NR: AP600	7832				· · · · · · · · · · · · · · · · · · ·		1	
function. The on the diffusifigures.	device ha	as been ments in	used for groups I	studying the o	effect whi zermanium.	ch dislocati Orig. art.	ons have has: 2	
SUB CODE: 20/	SUBM	DATE:	09Feb65/	ORIG REF	: 004/	OTH REF:	003	
					•			
	•						•	
				· · ·				
				:		±1		
							•	
1 •	· · · · · ,					·		
ard 3/3da					•			

ERTEVTAN, L. II.

Ertev san, L. N. and Kokina, E. B. - "Some observations on the biocheminal changes in the blood in hypertonia during its surgical treatment," Trudy Leningr. gospitalya dlya lecheniya invulidov Otechestv. voyny, Leningrad, 1948, p. 31/-24

SO: U3950, 16 June 53, (Letopis, 'Zhurnal 'nykh Statey, No. 5, 1949)

ERTEVISAN, L. H.

Ertevtsan, L. N. - "Some ovservations on the conditionof peripheral vascula in persons suffering from hypertonia during surgical treatment," Trudy Leningr. obl. gospitalya dlya lecheniya invalidov Otechestv. voyny, Leningrad, 1948, p. 355-59

SO: U-3950, 16 $^{\rm J}$ une 53, (Letopis, $^{\rm t}$ Zhurnal $^{\rm t}$ nykh Statey, No. 5, 1949).

ERTEVISIAN, L.N.

Third All-Russian Scientific Conference of Neurosurgeons. Vop.neurokhir. 17 no.2:59-63 Mr-Ap '53. (MLRA 6:5)

(Nervous system--Surgery)

ERTEVISIAN, L.W., starshiy nauchnyy sotrudnik.

Fourth All-Bassian Conference of Meurosurgeons on Theory and Practice. Vopr. neirokhir. 17 no.6:56-60 M-D '53. (MEMA 6:12) (Mervous system-Surgery)

ERTKYTSIAN, L.N.

Remote results of surgical treatment of hypertension. Vop.neiro-khir. 18 no.2:35-43 Mr-Ap '54. (MLRA 7:5)

1. Iz otdeleniya neyrososudistykh sabolevaniy Leningradskogo nauchno-issledovatel skogo instituta imeni prof. A.L. Polenova. (Postupila v redaktsiyu 23.IX.1953) (HYPERTENSIOE, surgery, *results, remote)

BONDARCHUK, A.V., professor; KRTEVTSIAN, L.N.; SAVCHKNKO, Yu.N.

First conference of Bashkir neurosurgeons. Vop.neirokhir. 20 no.3:
46-47 My-Je 156.

(BASHKIRIA--NERVOUS SYSTEM--SURGERY)

ESTEVISIAN LING

SHAMOV, V.N., sasluzhennyy deyatel nauki, professor; ERTEVISIAN, L.N., kandidat meditsinskikh nauk

Organization of neurosurgical care in province and city hospitals in the R.S.F.S.R. Sov.med. 21 no.4:143-147 Ap '57. (MIRA 10:7)

1. Glavnyy neyrokhirurg RSFSR, deystvitel'nyy chlen Akademii meditsinskikh nauk SSSR (for Shamov)
(NEUROSURGERY
in hosp. in Russia)

SHAMOV, V.N., zasluzhennyy deyatel' nauki prof.; ERTEVTSIAN, L.N.

Work of the Leningrad Neurosurgical Institute in training personnel.

Von.neirokhir. 22 no.2:52-55 M-Ap '58. (MIRA 11:4)

Vop.neirokhir. 22 no.2:52-55 M-Ap '58.
(BRAIM-SURGERY)
(MEDICINE-STUDY AND TEACHING)

Change in arterial pressure in hypertensive patients during

With and 26 no. 11.50.54 N 158 (MIRA 11:12)

Burgery. Klin.med. 36 no.11:50-54 N '58 (MIRA (HYPERTENSION, physiol. eff. of surg. on arterial pressure (Rus))

ERTEVISIAN, L.N.

Changes in the electrocardiogram in cats with a brain trauma and radiation sickness. Med. rad. 8 no.3:38-43 Mr '63. (MIRA 17:9)

1. Iz Leningradskogo neyrokhirurgicheskogo instituta imeni prof. A.L. Polenova (dir. - prof. V.M.Ugryumov).

ERTEVTSIAN, L.N., kand. med. nauk

Pathomorphological changes in the brain in cats following surgical injury. Vop. neirokhir. 27 no.4:52-55 Jl-Ag'63 (MIRA 17:2)

1. Leningradskiy neyrokhirurgicheskiy institut imeni A.L. Polenova (dir. - prof. V.M.Ugryumov).

ERTEVTSYAN, Ye.K.

Parasites of scale insects of greenhouse plants in the Erivan region. Izv. AN Arm. SSR. Biol. nauki 18 no.11: 99-104 N '65. (MIRA 19:1)

1. Zoologicheskiy institut AN Armyanskoy SSR. Submitted May 26, 1965.

ERTL,	Istvan						
	The use of	conveying	trundles.	Musz elet	16 no.20:12	¹61 .	
					·.		

ERTL, Istvan

Corrugations on rails. Musz elet 18 nc.26:10 19 D '63.

ERTL, Istvan

Street loadings in city transportation. Kozleked kozl 19 no.44:736-739 3 N 163.

ERTL, Istvan, foeloado

Using small-capacity containers in Hungarian transportation is 15 years old. Kozleked kozl 21 no.1:10-13 3 Ja '65.

1. Financial Department of the Ministry of Transportation and Postal Affairs, Budapest.

ERTL, J.

"The 1st Czechoslovak National Glider Contest" p. 12; "What does it mean?" p. 13; (REPULES, Vol. 6, No. 18, Sept. 1953, Budapest, Hungary)

SO: Monthly List of East European Accessions, L.C., Vol. 2, No. 11, Nov. 1953, Uncl.

ERTL, J.

"Gliders in Slovakia", P. 440, (KRIDLA VLASTI, Vol. 4, No. 19, Sept. 1954, Fraha, Czechoslovakia)

SO: Fonthly List of East European Accessions, (EEAL), IC, Vol. 4, No. 1, Jan. 1955, Uncl.

ERTL, J.

"Activity of Our Gliding Stations", P. 243, (KRIDLA VLASTI, No. 11, May 1954, Praha, Czechoslovakia)

SO: Monthly List of East European Accessions, (EEAL), LC, Vol. 3, No. 12, Dec. 1954, Uncl.

ERTL, Jozsef (Bratislava)

Bratislava is the site of the 1st Stunt Flight World Championship. Repules 13 no.5:10-13 My '60.

ERTL, Jozsef

The Czechoslovak Stunt Flying Championship. Repules 14 no.10:7 0 161.

1. A szlovak teruleti "Obranca vlasti" foszerkesztoje.

AROS, B.; ERTL, M.

Hypothalamic changes in renal hypertension. Acta morph. acad. sci. hung. 11 no.3:311-318 '62.

1. Institute of Histology and Embryology (Director: Prof. I. Toro),
Medical University, Budapest.
(HYPOTHALAMUS) (HYPERTENSION RENAL) (PARABIOSIS)
(VASOPRESSIN)

ERTL, Milan

The 6th Conference of Czechoslovak Hydrobiologists and Ichthyologists. Vestnik CSAV 71 no.1:88-89 *62.

ERTL, MILAN.

Rhizopoda raseliniska Bor (Orava). Bratislava, Vyd. Slovenskej akademie vied, 1955. 39 p. (Slovenska akademia vied. Sekcia 2. Prace. Seria biologicka, zv. 1, zosit 12) (Rhizopoda of the Bor peat bog in the district of Orava. German and Russian summaries. illus., bibl.)

SOURCE: East European Accessions List, (EEAL) Library of Congress, Vol. 5, No. 8, August 1956.

Critical reports on the slecies biffulnis leligi Whiles. p.23.

BICLOIA. (Slove ske ekederia vied) Bratislana. 'cl. 1, no. 7,
1056.

SC ROE: East Suropean Accessions List, (real), Library of Congress,
Yel. 5 no. 12, December 1956.

ERTL, M.

ERTL, M. Colonies of cladocera and copepoda on two Orava moors. p. 465.

Vol. 11, No. 8, 1956. RIOLOGIA SCIENCE Bratislava, Czechoslovakia

So: East European Accession, Vol. 6, No. 2, Feb. 1957

ERTL, M.; JURIS, S.

Plankton of astatic waters in the area of Orava Dam.

P. 352, (Biologia) Vol. 12, no. 5, 1957, Praha, Czechoslovakia

SO: Monthly Index of East European Acessions (EEAI) Vol. 6, No. 11 November 1957

ERTL, M: VHANOVURY, M

"List of Cladocers found on the Great Schutt, 1953-1957."

BIOLOGIA, Bratislava, Czechoslovakia, Vol. 13, no. 6, 1958

Monthly list of East Europe Accessions (EEAI), LC, Vol. 8, No. 6, Sept 59 Uncles

"APPROVED FOR RELEASE: Thursday, July 27, 2000

CIA-RDP86-00513R00041222

allel, milen SURTALE, Given Names

> Czechoslovakia Country:

Academic Degrees: C. Sc. /Candidate of Sciences/

Affiliation: Biological Institute SAV /Slovenska akademia ved; Slovak Academy Affiliation: Of Sciences (Biologicky ustav SAV), Bratislava

Bratislava, Hasa Veda, Vol VIII, No 7, 1961, pp 408-411. Source:

"Life in the Branches of the Danube River." Datas

GPO 981643

. ERTL, Milan; ERTLOVA, Eva; LAC, Jan; VRANOVSKY, Marian

Bibliographic survey of the hydrofauna of the Czechoslovak section of the Danube during the year 1918-1958. In German. Biologia 16 no.1:57-73 '61. (EEAI 10:7)

1. Biologisches Institut der Slowakischen Akademie der Wissenschaften, Abteilung der Zoologie, Bratislava.

(DANUBE RIVER) (CZECHOSLOVAKIA--FRESH-WATER FAUNA)

CZECHOSLOVAKIA

ERTL, Milan; Czechoslovak Academy of Sciences, Department of Hydrobiology at the Biological Institute of the Slovak Academy of Sciences / CSAV, Biologicky Ustav Slovenskej Akademie Vied, Oddelenie Hydrobiologie /, Bratislava.

"Contribution to the Information on Winter Zooplankton of Strbske Pleso ".

Eratislava, Biologia, Vol 18, No 10, 1963, pp 767 - 791

Abstract: The amount of the plankton and the distribution of varieties of which it consists are given. The samples were taken at the beginning of April 1963 when the water was covered with a 70 cm thick layer of ice. 10 cm below ice oxygen content was 6 mg/l at the bottom at 17 m no oxygen was found. pH was 6.7. Keratella hiemalis formed 85% of the plankton. 1 Figure, 3 Tables, 2 Czech, 4 Polish, 3 German, 2 Hungarian references.

1/1

L 59583-65 ACCESSION NR: AP5020536

cz/0049/64/000/009/0675/0689

AUTHOR: Ertl, Milan (Graduate biologist, Cundidate of sciences); Vrenovsky, Murian (Graduate biologist)

TITLE: Zooplankton of Poprad Hountain Lake

SOURCE: Biologia, no. 9, 1964, 675-689

TOPIC TAGS: biologic ecology, microbiology, surface water

ABSTRACT: There are only a limited number of species occurring in the lake. In pelagic waters it varied from 470/liter to 59/liter. The dominant organism throughout the year is Keratella hiemalis Carlin, which is most frequent at the time of the melting of the ice. Its proportion in the total number of organisms varied from 0.5 to 98.8%. In July 62, 105 organisms of Cyclops tatricus Kozminski were found in 1 liter of water. Orig. art. has: 4 figures, 6 tables.

ASSOCIATION: Biologicky ustaw Slovenskej akademie vied, Oddelenie hydrobiologie, Bratislava (Hydrobiology Department, Biological Institute of the Slovak Academy of Sciences)

SUBMITTED: 12Mar64 NR REF SOV; 001

Card 1/1 //

ENCL: 00 OTHER: 018 SUB CODE: LS, ES

JPRS

Hydrobiology

CZECHOSLOVAKIA

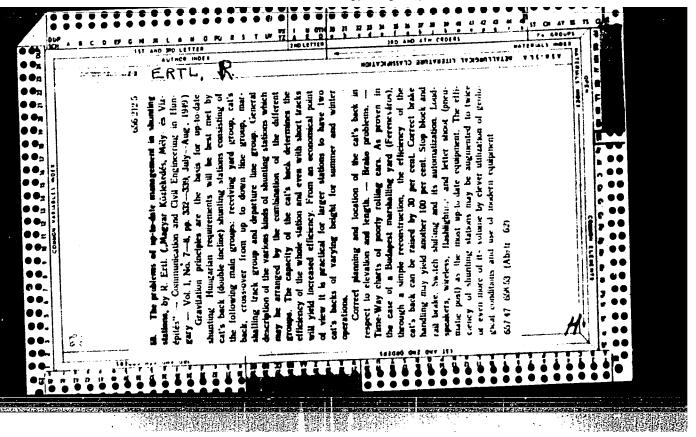
ERTL, Milan: Institute of Biology of the Countryside, Slovak Academy of Sciences, Department of Hydrobiology (Ustav Biologie Krajiny Slovenskej Akademie Vied, Oddelenie Hydrobiologie), Bratislava.

"Zooplankton of the Czechoslovak Section of the Danube."

Bratislava, Biologia, Vol 21, No 7, 1966, pp 545 - 548

Abstract: The amount of zooplankton in the investigated part of the Danube is not too great because of the high flow velocity of the water. Rotifera form the greatest part of the zooplankton. In the nannozooplankton flagellata are found most frequently; they occur mostly in the fall. Ciliata are rather rare. Small amoebas are found in substantial numbers because of the high seston content. 7 Czech, 2 East German roferences. (Manuscript received 24 May 65).

1/1



ERTL, R:

TECHNOLOGY

K MILEKEDESTUDOMANYI SZEMLE. (Kozlekedes- es Kozlekedesepitestudomanyi Egyesulet) Budapest.

ERTL, R: VARGA, I. The Pioneer Railroad of the Hungarian State Railways to Mount Szechenyi is 10 years old. p. 314.

Vol. 8, no. 7/8, July/Aug. 1958.

Monthly List of East European Accession (EEA1) LCV lo. 8, NO. 3
March 1959, Unclass.

ERTL, R.

Possibilities and advantages of introducing reversing-shuttle train sets in the passenger traffic of the dungarian State Railways in the area of Eudapest. p. 148

KOZIEKEDEST'TOMANYI, SZEELE. (Kozlekedes- es Kozlekedesepitestudomanyi Egyesulet) Budapest, Hungary, Vol. 9, no. 4, Apr. 1959.

Monthly list of East European Accessions (FEAI), LC, Vol. 8, No. 8, August 1959.
Uncla.

The Southern Railroad Station is being rebuilt. Magy ep ipar 11 no.6: 241-245 62.

ERTL, Robert, dr., a muszaki tudomanyok kandidatusa

Reconstruction of the Budapest Southern Railroad Station. Kozl tud sz 12 no.9:385-392 S 162.

1. Magyar Allamvasutak Vasuttervezo U.V. fomernoke, es "Kozleke-destudomanyi Szemle" szerkeszto bizottsagi tagja.

ERTL, but rt, ar.

Recessity of building Island platforms. Vasut 14 no.7:14-15 S 164.

1. Railroad Planning Enterprise, Hungarian Stato hailways, Endapest.

BINA, Jaroslav, ERTL, Tibor

Radiation vulcanization of silicone rubber. Jaderna energie 9 no.4:125-127 Ap '63.

1. Vyskumny ustav kablov a izolantov, Bratislava.

ERTL, Tobert, dr., a muszaki tudomanyok kandidatusa, fomernok

Modernization of railroad stations with island platforms. Kozl tud sz 13 no.8:353-359 Ag 63

1. MAV Vasuttervezo U.V.; "Kozlekedestudomanyi Szemle" szerkeszto bizottsagi tagja.

CZECHOSLOVAKIA / General Biology - General Hydrobiology.

Abs Jour: Ref Zhur-Biol., No 9, 1958, 38104.

Author : Ertl, Yurish.

: Not given. Inst

: Plankton of Astatic Waters in the Vicinity of Title

the Orav Reservoir.

Orig Pub: Biologia, 1957, 12, No 5, 352-361.

Abstract: A study was conducted of plankton in shallow pools

which settled on the territory of the Orav Reser-

voir after partial water draining. Lists are given of phytoplankton and zooplankton organisms. Two pool groups are listed (on mineral substrate and on peat moss), which differ in depth, pH value, and composition of zooplankton species. It seems that after the reservoir is refilled, the peat moss exerts no effect on the quality of

the water.

Card 1/1

ERTLI, A.A.

Atelectasis following surgery in pulmonary tuberculosis. Probl. tub. 41. no.3:40-45'63. (MIRA 16:9)

1. Iz 1-go legoshnokhirurgicheskogo otdeleniya (zav. -kand. med. nauk P.A.Semenkin) Moskovskogo nauchno-issledovatel!-skogo instituta tuberkuleza (dir. - kand. med. nauk T.P. Mochalova, zamestitel! direktora po nauchnoy chasti - prof. D.D.Aseyev) Ministerstva zdravookhraneniya RSFSR. (LUNGS—COLLAPSE) (LUNGS—SURGERY)

SEMENKIN, P.A.; ERTLI, A.A.; KARAMY CHEVA, V.N.

Surgical treatment of children and adolescents with join andy tuberculosis. Probl. tub. 42 no.8324-28 164.

 $(Mik_{\perp}/2/11)$

I. 1-ye legochno-khirurgicheskoye otdeleniya (rav. - kand.ned. nauk P.A.Semenkin) Moskovrkogo randino-tanjanavatel (alee i instituta turerkuleza (direktor - kand.met.rank T.P.Movis lova: zamestitel direktora po nauchnoy chasti prof. D.D.Aseyev) Ministerstva zdravookhraneniya MOFSR.

ERTL, Milan; ERTLOVA, Eya; LAC, Jan; VRANOVSKY, Marian

Bibliographic survey of the hydrofauna of the Czechoslovak section of the Danube during the year 1918-1958. In German. Biologia 16 no.1:57-73 161. (EEAI 10:7)

1. Biologisches Institut der Slowakischen Akademie der Wissenschaften, Abteilung der Zoologie, Bratislava.
(DANUBE RIVER) (CZECHOSLOVAKIA--FRESH-WATER FAUNA)

ERTLOVA, Eva: Biology Institute (Biologicky institut) of the Slovak Academy of Sciences, Bratislava, CZECHOSLOVAKIA.

"Chironomidae of the Danube."

Bratislava, Biologia, Vol 18, No 8, 1963; pp 612-620.

Abstract [German article]: Ecologic study of larvae of Danubian zoobenthic species: 2 of Tanypodinae, 9 of Orthocladicinae, 12 of Chironomariae and 5 of Tanytarsariae are described. Table shows habitat (sand, river bottom, free-swimming, etc.). One German, 1 Bulgarian, 5 Soviet and 7 Czech references.

1/1

L 59582-65

ACCESSION NR: AP5020535

cz/0049/64/000/009/0666/0674

AUTHOR: Ertlova, Eva (Ertlova, Yeva) (Graduate biologist)

3 B

TITLE: Contribution to the knowledge about zoobenthos of Poprad Mountain Lake

SOURCE: Biologia, no. 9, 1964, 666-674

TOPIC TAGS: biologic ecology, entomology, microbiology, surface water

ABSTRACT: The zoobenthos in the lake is quite developed; the most frequent form are worms, namely Peloscolex ferox. Next come larvae of mosquitos namely Eutanytarsus gr. gregarius, Procladius sp. and Psectrocladius gr. psilopterus. Among the Chironomidae the most frequently found are Prodiamesa gr. olivacea and Eutanytarsus gr. gregarius. In September 62 and April 63 larvae of genus Sergentia were found. The greatest thickness of zoobenthos is found in the deepest part of the lake; 11,381 benthos per sq. meter were found in Apr. 63, 95.9% of which were Oligochaeta.

C-1 1/2

L 59582 - 65		
ACCESSION NR: AP5020535		Ó
Near the shore the maximum (Certain differences were for by HRABE in 1942, and the fand Trichotomesa found by b	und between the Rings of the indines of the	ropelopia
A	Landing Control of the Control of th	
Orig. art. has: 1 figure, 5 tabl	165 • 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1.	
ASSOCIATION: Biologicky ustav Sl Bratislava (Department of Hydrobi Academy of Sciences)	lovenskej akademie viud, Oldelen	ie hydrobiologie. the Slovak
ASSOCIATION: Biologicky ustav Sl Bratislava (Department of Hydrobi	lovenskej akademie viud, Oldelen	ie hydrobiologie; the Slovak SUB CODE: LS.ES
ASSOCIATION: Biologicky ustav Sl Bratislava (Department of Hydrobi Academy of Sciences) SUBMITTED: 12Mar64	lovenskej skademie viud, Oddelen lology, Biological Institute of	the Slovak
ASSOCIATION: Biologicky ustav Sl Bratislava (Department of Hydrobi Academy of Sciences) SUBMITTED: 12Mar64	lovenskej skademie viud, Oddelen lology, Biological Institute of ENCL:)OO	SUB CODE: LS,ES
ASSOCIATION: Biologicky ustav Sl Bratislava (Department of Hydrobi Academy of Sciences)	lovenskej skademie viud, Oddelen lology, Biological Institute of ENCL:)OO	SUB CODE: LS,ES

ERTSEGOVATS, VASICH
YUGOSLAVIA/Virology - Human and Animal Viruses.

E-3

Abs Jour : Ref Zhur - Biol., No 3, 1958, 9686

Author : Ertsegovats, Vasich

Inst

Title : A Study of Duration of Immunity Created After Vaccination

Against Rabies by Khempt's method.

: Acta veterin., 1956, 6, 3-4, 54-56 Orig Pub

Abstract : The vaccine injected hypodermically into rabbits at a

> dose of 10 ml protected them for a period of 8 months from development of disease, brought about by introduc-

tion of 2 ml of the street virus strain.

Card 1/1

ERTSEY, Csaba, dr.; KOVALKOVITS, Istvan, dr.

Hernia obturatoria. Magy. sebesz. 17 no.1:37-40 F.64.

1. Hevesmegyei Tanacs Korhaza, Eger. I - II. sz. Sebeszeti Osztaly.

ERTUGANOVA, Z.A.

Development of inflammatory reaction in various functional states of the central nervous system. Biul.eksp.biol.i med. 37 no.2: 32-34 F 154. (MIRA 7:6)

ERTUGANOVA, Z.A.

Effect of denervation on the development of necrosis following intracutaneous administration of staphylococcal culture. Biul. eksp. biol. i med. 38 no.9:24-26 S '54. (MLRA 7:12)

1. Iz otdela infektsionnoy patologii i eksperimental'noy terapii (zav. prof. Kh.Kh.Planel'es) Instituta epidemiologii i mikrobiologii imeni N.F.Gamalei (dir. deystvitel'nyy chlen AMN SSSR prof. V.D.Timakov) AMN SSSR, Moskva.)

(NERVOUS SYSTEM, physiology.

eff. of denervation of organ on develop. of necrosis after intracutaneous admin. of Micrococcus pyogenes in rabbit)

(MICROCOCCAL INFECTIONS, experimental,

pyogenes, eff. of denervation of organ on develop of necrosis after intracutaneous admin. of Micrococcus pyogenes culture in rabbit)

(NECROSIS, experimental,

eff. of denervation of organ on develop. of necrosis after intracutaneous admin. of Micrococcus pyogenes culture in rabbit)

"APPROVED FOR RELEASE: Thursday, July 27, 2000

CIA-RDP86-00513R00041222

ERTUGANOVA, Z.A.

USSR/Medicine - Pneumococci, Antibiotics, Antihistaminics

PD-2316

Card 1/1

Pub 148 - 17/36

Author

: Ertuganova, Z. A.

Title

: The combined action of antibiotics and antihistaminics in pacumo-

cocci infections

Periodical

: Zhur. mikro. epid. i immun. No 2, 49-51, Feb 1955

Abstract

: Treatment of experimental pneumococcal infections of rabbits indicated that penicillin and dimedral reinforce each other's effect. Penicillin and dimedrol exert a similar action: the first suppresses the formation of toxic substances by the bacteria, while the second, being an antihistaminic, counteracts phlogogenic substances, particularly histamine, which produce infarmation. Six

graphs.

Institution : Division of Infection Pathology and Experimental Therapy, Insti-

tute of Epidemiology and Microbiology imeni N. F. Gamaleya,

Academy Medical Sciences USSR

Submitted

: December 12, 1953

ERTUGANOVA, Z.A., kandidat meditsinskikh nauk

Cure for nettle rash. Neuka i shism' 22 no.9:64 S'55. (MLRA 8:12) (Urticaria)

ERTUGANOVA, Z. A.

"The Influence of Antihisteamine Substances on the Development of an Inflomatory Process." Proceedings Inst. Epidem. and Microbiol. im. Campleya, 1954-56.

Division of Infectious Pathology and Experimental Therapy Planel'yes, Kh. Kh., Doc of Ned Sci, Corr Nem of Acad of Med Sci USSR, head. Inst. Epidem and Microbiol. im Camaleyn, AMS USSR.

SO: Sum 1186, 11 Jan 57.

ERTUGANOVA, Z.A. (Cand. of Med. Sci.); KALININA, N.A.; KIVMAN, G. Ya. (Cand. of Med. Sci.); BREGER, M.A. (Cand. of Med. Sci.); IVANOVA, G.A. (Cand. of Vet. Sci.); BALYN', I.R.;

"Tetracyclin,"

p. 214 Ministry of Health USSR Proceedings of the Second All-Union Conference on Antibiotics, 31 May - 9 June 1957. p. 405, Moscow, Medgiz, 1957.